

Journal of The Chemical Society, Chemical Communications

NUMBER 8/1974

17 APRIL

Tenellin and Bassianin, Metabolites of *Beauveria* Species. Structure Elucidation with ^{15}N - and Doubly ^{13}C -Enriched Compounds using ^{13}C Nuclear Magnetic Resonance Spectroscopy

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Summary ^{13}C n.m.r. spectroscopy and biosynthetic labeling with ^{15}N nitrate and $[1,2-^{13}\text{C}]$ acetate proved to be valuable adjuncts to established chemical and spectroscopic methods in elucidating the structures of tenellin and bassianin as 3-acyl derivatives of 1,4-dihydroxy-5-*p*-hydroxyphenyl)-2(1H)-pyridone.

A PREVIOUS publication¹ contains preliminary u.v., i.r., and mass spectral data on two related pigments tenellin (I) ($\text{C}_{21}\text{H}_{23}\text{NO}_5$), and bassianin ($\text{C}_{23}\text{H}_{25}\text{NO}_5$), as well as conditions for their production and isolation from the insect pathogenic fungi *Beauveria tenella* (Delacroix) Siem., and *Beauveria bassiana* (Bals.) Vuill. We now report their structures.

Methylation of (I) with Ag_2O -MeI progressively methylates the hydroxamic acid [(II); m.p. 194°], phenol [(III); m.p. 96–98°], and enol [(IV); wax] hydroxy-groups. High resolution mass spectral studies, including precursor ion experiments, showed a major fragmentation pathway *via* initial loss of oxygen in (I), whereas in contrast fragmentation of (II), (III), and (IV) involved initial loss of CH_2O . This provides strong evidence for a cyclic hydroxamic acid unit² in (I).

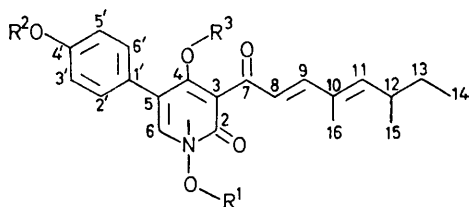
Compound (I), $[\alpha]_D^{25} - 44.0^\circ$ (1% in acetone) with alkaline peroxide gave *p*-hydroxybenzoic acid; in refluxing KOH by retro-aldol cleavage it gave the 3-acetyl derivative

(V), m.p. 230–231° ($\text{C}_{13}\text{H}_{11}\text{NO}_5$, *m/e* 261-0635), and two aldehydes, both racemic mixtures, characterized as the 2,4-dinitrophenylhydrazones of 2-methylbutanal, m.p.³ 128.5° and 2,4-dimethylhex-2-enal, m.p.⁴ 160–161°.

The ^1H n.m.r. signals [$(\text{CD}_3)_2\text{SO}$; 100 MHz] of (I) could be assigned as follows: δ 17.0br [s, 1H, $-\text{C}(\text{OH})=\text{C}(\text{C}=\text{O})-$], 11.5br (s, 1H, $>\text{NOH}$), 9.5br (s, 1H, ArOH; this and previous OH's temperature dependent, exchanged with D_2O), 8.10 (s, 1H, 6-H), 7.78 [AB, 2H, 8- and 9-H, $\Delta\nu_{\text{AB}}$ 50.4, J_{AB} 15.3 Hz, 9-H long range coupled <0.5 Hz to 11-H (double resonance)], 7.07 (AA'BB', 4H, $\Delta\nu_{\text{AB}}$ 49.6 Hz, J_{AB} 8.6 Hz, 1,4-subst. Ar), 5.92br [d, 1H, 11-H, $J_{11,12}$ 9.6 Hz, coupled <0.3 Hz to vinylic Me (double resonance)], 2.50 (m, 1H, 12-H, Bu^8 - $>\text{CH}$ -), 1.88br (s, 3H, 16-H, vinylic Me), 1.38 (m, 2H, 13-H, Bu^8 - $>\text{CH}_2$), 1.01 (d, 3H, 15-H, Bu^8 -Me, $J_{12,15}$ 6.3 Hz), and 0.85br (t, 3H, 14-H, remaining Bu^8 -Me). The only significant differences in the spectra of derivatives were: (II), δ 11.5 (OH) replaced by δ 3.98 ($>\text{NOMe}$); (III), δ 11.5 and 9.5 (OH) replaced by δ 4.02 ($>\text{NOMe}$) and 3.79 (ArOMe); (V), all signals for H-8 replaced by acetyl Me at δ 2.70. Structural, stereochemical, and conformational features of (I) are thus retained in (II) and (III); (V) differs only in the contraction of an acyl to an acetyl substituent. A nuclear Overhauser enhancement (NOE) study⁵ [irradiated proton(s) in brackets] of (II) gave: 9-H {11-H} 22%; 8-H {11-H} 0%; 9-H {allylic Me} 0%; 8-H

{allylic Me} 20%; 6-H {>NOMe} 8%; 6-H {2'- and 6'-H} 26%; 3'- and 5'-H {2'- and 6'-H} 10%.

The preceding evidence requires that (I), (II), and (III) possess 4,6-dimethylocta-*trans,trans*-2,4-dienoyl and HOC₆-H₄C units, with 6-H close to the chemically equivalent aromatic hydrogens 2'-H and 6'-H, as well as to the *N*-methoxy-group in (II) and (III), and hence the *N*-hydroxy-group in (I).



(I) $R^1 = R^2 = R^3 = H$

(II) $R^1 = Me, R^2 = R^3 = H$

(III) $R^1 = R^2 = Me, R^3 = H$

(IV) $R^1 = R^2 = R^3 = Me$

Pulse Fourier transform (8K; data accuracy ± 0.6 Hz) ^{13}C n.m.r. spectral data [(CD₃)₂SO; internal reference Me₄Si, 5 kHz sweep width, 25.16 MHz] were obtained from proton-noise decoupled (p.n.d.) and high resolution spectra, and assignments were aided by off-resonance and single 1H frequency decoupling experiments.^{6,7} Assignments for (I) were: δ 193.8 (C-7), 173.0 (C-4), 157.5 (C-2), 156.9 (C-4'), 150.8 (C-11, $^1J_{CH}$ 154.4 Hz), 149.7 (C-9, $^1J_{CH}$ 155.9 Hz), 140.0 (C-6, $^1J_{CH}$ 183.6 Hz), 132.6 (C-10), 130.2 (C-2', -6', $^1J_{CH}$ 160.0 Hz), 123.1 (C-8, $^1J_{CH}$ 167.5 Hz, 122.8 (C-1'), 115.0 (C-3', -5', $^1J_{CH}$ 160.0 Hz), 110.9 (C-5), 105.9 (C-3), 34.6 (C-12, $^1J_{CH}$ 125.4 Hz), 29.4 (C-13, $^1J_{CH}$ 126.0 Hz), 19.8 (C-15, $^1J_{CH}$ 125.8 Hz), 12.3 (C-16, $^1J_{CH}$ 126.2 Hz), 11.7 (C-14, $^1J_{CH}$ 125.2 Hz).

Addition of 90% enriched [1,2- ^{13}C] acetate to cultures of *B. bassiana* yielded (I) containing five discrete two-carbon units each doubly labelled with ^{13}C . The presence of ^{13}C - ^{13}C spin-spin coupling in the p.n.d. spectrum proved the existence of the following pairs of directly bonded carbons: C-2, C-3 ($^1J_{CC}$ 75.3 Hz); C-7, C-8 ($^1J_{CC}$ 55.9 Hz); C-9, C-10 ($^1J_{CC}$ 53.6 Hz); C-11, C-12 ($^1J_{CC}$ 43.2 Hz); C-13, C-14 ($^1J_{CC}$ 34.4 Hz).

In addition the p.n.d. spectrum of (I) 95% enriched with ^{15}N (obtained from cultures with K $^{15}NO_3$ as the only nitrogen source) established that only C-6 [$^1J(^{15}NC)$ 15.0 Hz], C-2 [$^1J(^{15}NC)$ 11.0 Hz], and C-3 [$^2J(^{15}NC)$ 9.2 Hz] were spin-spin coupled to ^{15}N , and only H-6 [$^2J(^{15}NH)$ 1.0 Hz] was similarly coupled to nitrogen in the corresponding 1H n.m.r. spectrum. These data confirm that the nitrogen in (I) is geminal to 6-H and C-3, and directly bonded to C-6 and C-2;⁸ the latter observation provides further evidence for a hydroxamic acid group.

The combined results therefore require that tenellin possesses a 1-hydroxy-2-pyridone ring substituted as shown in structure (I). Our NOE study places H-6 and the *p*-hydroxyphenyl substituent on adjacent carbons. C-4, C-3, and C-7 must constitute the enol form of a β -diketone system in order to account for their chemical shift,^{7a} and the presence of the strongly hydrogen bonded hydroxy-group observed in the 1H n.m.r. spectrum; the chemical shift of C-5 is consistent with it being deshielded by the aryl group and shielded by β -nitrogen and hydroxy substituents. Tenellin is therefore 3-(4,6-dimethyl-*trans-trans*-octa-2,4-dienoyl)-1,4-dihydroxy-5-(*p*-hydroxyphenyl)-2(1*H*)-pyridone.

Parallel studies on bassianin gave a structure differing only in that the C-3 acyl substituent is extended by an additional conjugated *trans*-substituted -CH=CH- unit positioned between C-7 and C-8 of the structure (I).

(Received, 17th December 1973; Com. 1696.)

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